BOROKSİT
SAFETY DATA SHEET

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SECTION 1. Identification of the Substance and the Company

1.1. Product identifier

Substance name : Diboron trioxide

Trade name : BOROKSİT

Chemical name/synonyms: Boron trioxide, diboron trioxide, anhydrous boric acid, boric oxide

Index N° : 005-008-00-8

CAS N° : 1303-86-2

EC N° : 215-125-8

REACH Registration number : 01-2119486655-24-0003

1.2. Relevant identified uses of the substance and uses advised against

Relevant identified uses

The product is used in industrial manufacturing, among others in:

- Ceramics
- Detergent
- Borosilicate glass
- Textile fibreglass

For area-specific use, see the exposure scenarios in the annex of this extended Safety Data Sheet (eSDS).

Uses advised against

Not applicable, there are no uses of Boroksit advised against.

1.3. Details of the supplier of the safety data sheet

Importer

Name : AB ETIPRODUCTS OY

Address **:** Piispanportti 5, 02240 Espoo/FINLAND

Phone No **:** + 358 9 819 444 40

Fax No **:** + 358 9 819 444 44

e-mail **:** sales@etiproducts.com

Manufacturer

Name : ETİ MADEN İŞLETMELERİ GENEL MÜDÜRLÜĞÜ

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Phone No : +90 312 294 20 00

Fax No : +90 312 232 71 84

1.4. Emergency phone number: +49 (0)6132-84463 (24-Hour-Number) GBK GmbH

SECTION 2. Hazard Identification

2.1. Classification of the substance

2.1.1. Classification according to Regulation (EC) N°1272/2008 (CLP)

Harmonized classification provided in the 1st ATP to CLP (Regulation EC N° 790/2009)

Repr. Cat. 1B; H360FD

Specific concentrations limits : Repr. 1B; H360FD: C ≥3.1%

Precautionary Statement Prevention : P201; P202; P280

Precautionary Statement Response : P308+P313

Precautionary Statement Storage : P405

Precautionary Statement Disposal : P501

2.1.2. Additional information

For the full text of Hazard Class/Statements and Precautionary Statements see SECTION 16.3.

2.2. Label elements

2.2.1. Label according to Regulation (EC) N°1272/2008 (CLP)

Hazard pictograms

Signal word **:** Danger

Hazard Statements **:** H360FD: May damage fertility or the unborn child.

Precautionary Statements:

P201 : Obtain special instruction before use.

P202 : Do not handle until all safety precautions have been read and understood.

P280 : Wear protective gloves/protective clothing/eye protection/face protection.

P308+P313 : IF exposed or concerned: Get medical advice/attention.

P405 : Store locked up.

2.2.2. According to REACH, Annex XVII

Restricted to professional users

2.3. Other hazards

Emergency overview

Boroksit is a white odourless, solid (glassy/porous) substance that is not flammable, combustible or explosive, and has low acute oral and dermal toxicity.

Potential health effects

Inhalation is the most significant route of exposure in occupational and other settings. Dermal exposure is not usually a concern because Boroksit is poorly absorbed through intact skin.

Inhalation

Occasional mild irritation effects to nose and throat may occur from inhalation of Boroksit dusts at levels higher than 10 mg/m^3.

Eye contact

Boroksit is non-irritating to eyes in normal industrial use.

Skin contact

Boroksit does not cause irritation to intact skin.

Ingestion

Products containing Boroksit are not intended for ingestion. Boroksit has low acute toxicity. Small amounts (e.g. a teaspoon) swallowed accidentally are not likely to cause effects; swallowing amounts larger than that may cause gastrointestinal symptoms.

Reproductive/developmental

Animal ingestion studies in several species, at high doses, indicate that borates cause reproductive and developmental effects [1]. A human study of occupational exposure to borate dust showed no adverse effect on reproduction. An epidemiological study and a peer reviewing report of the past epidemiological studies conducted in China didn't show any negative effect of boron on human fertility [2]. A study conducted in Turkey with boron exposed mine workers showed that mean blood concentrations of the high exposure group is ~6 times and ~9 times lower than those of the highest no effect level of boron in blood with regard to developmental and reprotoxic effects (respectively) in rats. With those findings, no unfavourable effects of boron exposure on reproductive indicators are observed in humans [3, 4].

Potential ecological effects

Large amounts of Boroksit can be harmful to plants and other species. Therefore releases to the environment should be minimized.

Signs and symptoms of exposure

Symptoms of accidental over-exposure to Boroksit have been associated with ingestion or absorption through large areas of damaged skin. These may include nausea, vomiting, and diarrhoea, with delayed effects of skin redness and peeling (see SECTION 11).

SECTION 3. Composition / Information on Ingredients

3.1. Substances

The product contains greater than 98.0 percent (%) Boroksit (B_2O_3).

Identification Name	EC N°	CAS N°	REACH Registration Number	Wt. %
Boron Oxide (Boron trioxide, diboron trioxide, anhydrous boric acid, boric oxide)	215-125-8	1303-86-2	01-2119486655-24-0003	> 98.0

For other "Chemical inventory listing", please refer to SECTION 15.

SECTION 4. First aid measures

4.1. Description of first aid measures

Skin contact

No treatment necessary because Boroksit does not cause irritation to intact skin.

Eye contact

No treatment necessary because non-irritant.

Inhalation

If symptoms such as nose or throat irritation are observed, remove person to fresh air. Boroksit has low inhalation toxicity.

Ingestion

If large amounts are swallowed (i.e. more than one teaspoon), contact a doctor or toxicity centre immediately.

4.2. Most important symptoms and effects, both acute and delayed

N.A.

4.3. Indication of any immediate medical attention and special treatment needed

Observation only is required for adult ingestion of less than 4 grams of Boroksit. For ingestion in excess of 4 grams, maintain adequate kidney function and force fluids. Gastric lavage is recommended for symptomatic patients only. Haemodialysis should be reserved for massive acute ingestion or patients with renal failure. Boron analyses of urine or blood are only useful for documenting exposure and should not be used to evaluate severity of poisoning or to guide treatment [5] (see SECTION 11).

SECTION 5. Firefighting measures

5.1. Extinguishing media

Any appropriate fire extinguishing media may be used on nearby fires.

5.2. Special hazards arising from the substance

Boroksit is not flammable, combustible or explosive. The product is itself a flame retardant.

5.3. Advise for firefighters

N.A.

SECTION 6. Accidental release measures

6.1. Personal precautions, protective equipment and emergency procedures

Avoid dust formation. In case of exposure to high level of airborne dust, wear a personal respirator in compliance with national legislation.

6.2. Environmental precautions

Boroksit is a water-soluble white product that may cause damage to trees or vegetation by root absorption (see SECTION 12).

6.3. Methods and material for containment and cleaning up

Land spill

Vacuum, shovel or sweep up Boroksit and place in containers for disposal in accordance with applicable local regulations. Avoid contamination of water bodies during clean up and disposal. No personal protective equipment is needed to clean up land spills.

Spillage into water

Where possible, remove any intact containers from the water. Advise local water authority that none of the affected water should be used for irrigation or for the abstraction of potable water until natural dilution returns the boron value to its normal environmental background level (see SECTIONS 12, 13 and 15).

6.4. Reference to other sections

See SECTIONS 8 and 13 for further information.

SECTION 7. Handling and Storage

7.1. Precautions for safe handling

To maintain package integrity and to minimise caking of the product, bags should be handled on a first-in first-out basis. Good housekeeping and dust prevention procedures should be followed to minimise dust generation and accumulation. Your supplier can advise you on safe handling, please contact the supplier.

7.2. Conditions for safe storage, including any incompatibilities

Dry, indoor storage is recommended since the product is highly hygroscopic. The product should be kept away from strong reducing agents.

7.3. Specific end uses

See exposure scenario in Annex to the SDS.

SECTION 8. Exposure controls / Personal protection

8.1. Control parameters

Occupational Exposure Limit Values

Substance:	**Boron oxide / Diboron trioxide**			
CAS N°	**1303-86-2**			
	Limit value-Eight hours		Limit value – Short term	
	ppm	mg/m^3	ppm	mg/m^3
Belgium		10		
Denmark		10		20
France		10		
Latvia		5		
Poland		10		
Spain		10		
Switzerland		10 inhalable aerosol		
United Kingdom		10		20

Source: IFA Institut für Arbeitsshutz der Deutschen Gesetzlichen Unfallversicherung

Occupational exposure limits for dust (total and respirable) are treated by OSHA, Cal OSHA and ACGIH as "Particulate Not Otherwise Classified" or "Nuisance Dust"

ACGIH/TLV : 10 mg/m^3

CAL OSHA/PEL : 10 mg/m^3

OSHA/PEL (total dust) : 15 mg/m^3

OSHA/PEL (respirable dust) : 5 mg/m^3

DNEL values

Exposure pattern	**Type/site of effect**	**Exposure route**	**DNEL value (for Boron)**
DNELs for workers			
Long-term	Systemic	Inhalation	1.45 mg BA/m^3
Long-term	Systemic	Dermal	4800 mg BA/day
DNELs for the general public			
Acute	Systemic	Oral	0.17 mg BA/kg bw/day
Long-term	Systemic	Dermal (external)	34.3 mg BA/kg bw/day
Long-term	Systemic	Dermal (systemic)	0.17 mg BA/kg bw/day
Long-term	Systemic	Inhalation	0.73 mg BA/m^3
Long-term	Systemic	Oral	0.17 mg BA/kg bw/day

Source: Chemical Safety Report of Boron Oxide

PNEC values

PNEC $_{\text{add, freshwater, marine water}}$ = 1.35 mg B/L

PNEC $_{\text{add aqua intermittent}}$ = 9.1 mg B/L

PNEC $_{\text{add freshwater sediment, marine water sediment}}$ = 1.8 mg B/kg sediment dry weight

PNEC $_{\text{add, STP}}$ = 1.75 mg B/L

Source: Chemical Safety Report of Boron Oxide

8.2. Exposure controls

8.2.1. Appropriate engineering controls

Maintain air concentrations below occupational exposure standards.

Use local exhaust ventilation to keep airborne concentrations of Boroksit dust below permissible exposure levels. Wash hands before breaks and at the end of the workday. Remove and wash soiled clothing.

8.2.2. Individual protection measures, such as personal protective equipment

Individual protection measures should be taken into account the Council Directive 89/966/EEC and the appropriate CEN standard.

Respiratory protection

In case of prolonged exposure to dust wear a personal respirator in compliance with national/international legislation (CEN standard).

Eyes and hands protection

Goggles and gloves are not required for normal industrial exposures, but may be warranted if environment is excessively dusty.

8.2.3. Environmental exposure controls

No special requirement.

SECTION 9. Physical and chemical properties

9.1. Information on basic physical and chemical properties

Appearance	: White solid, glassy/porous
Odour	: Odourless
Odour threshold	: N.A.
pH @ 20°C	: 4.4 (1.0% solution)
Melting point	: 450°C
Initial boiling point and boiling range	: 1860°C
Flash point	: Non flammable
Evaporation rate	: N.A.
Flammability (solid, gas)	: N.A.
Upper/lower flammability or explosive limits	: N.A.
Vapour pressure	: Negligible @ 20°C
Vapour density	: N.A.
Relative density	: N.A.
Solubility in water	: 2.7 % @ 20°C
Partition coefficient: n-octanol/water	: N.A.
Auto-ignition temperature	: N.A.
Decomposition temperature	: N.A.
Viscosity	: N.A.
Explosive properties	: Non explosive
Oxidising properties	: N.A.

9.2. Other information

Molecular weight	: 69.6
Specific gravity	: 2.04 @ 20°C

SECTION 10. Stability and reactivity

10.1. Reactivity

Boroksit is a stable product.

10.2. Chemical stability

Boroksit is a stable but hygroscopic product which absorbs moisture from the air. If moisture is present, Boroksit may cause corrosion of base metals.

10.3. Possibility of hazardous reactions

Reaction with strong reducing agents such as metal hydrides or alkali metals will generate hydrogen gas which could create an explosive hazard.

10.4. Conditions to avoid

Avoid contact with strong reducing agents.

10.5. Incompatible materials

Avoid contact with strong reducing agents such as metal hydrides or alkali metals.

10.6. Hazardous decomposition products

N.A.

SECTION 11. Toxicological information

11.1. Information on toxicological effect

11.1.1. Substances

Acute toxicity

Low acute oral toxicity; LD_{50} in rats > 2,600 mg/kg of body weight (Test material: Boron oxide) [6].

Skin corrosion / irritation

Low acute dermal toxicity; LD_{50} in rabbits is greater than 2,000 mg/kg of body weight [7]. Boron oxide is poorly absorbed through intact skin. Non-irritant.

Serious eye damage/ irritation

Boron oxide has no eye damage/irritation.

Respiratory or skin sensitisation

Boron oxide has no respiratory or skin sensitization.

Germ cell mutagenicity

Boron oxide is not mutagenic.

Carcinogenicity

Boron oxide is not carcinogenic.

Reproductive toxicity

Animal feeding studies in rat, mouse and dog, at high doses, have demonstrated effects on fertility and testes [1]. Studies in rat, mouse and rabbit, at high doses, demonstrate developmental effects on the foetus including foetal weight loss and minor skeletal variations. The doses administered were many times in excess of those which humans would normally be exposed to [8, 9]. While boron has been shown to adversely affect male reproduction in laboratory animals, there is no clear evidence of male reproductive effects attributable to boron in studies of

highly exposed workers. Human epidemiological studies show no increase in pulmonary disease in occupational populations with chronic exposures to borate dusts. An epidemiology study under the conditions of normal occupational exposure to borate dusts indicated no effect on fertility [2]. A study conducted in Turkey with boron exposed mine workers showed that mean blood concentrations of the high exposure group is ~6 times and ~9 times lower than those of the highest no effect level of boron in blood with regard to developmental and reprotoxic effects (respectively) in rats. With those findings, no unfavourable effects of boron exposure on reproductive indicators are observed in humans [3, 4].

STOT-single exposure

N.A.

STOT-repeated exposure

N.A.

Aspiration hazard

Boron oxide has no aspiration hazard.

SECTION 12. Ecological information

12.1. Toxicity

Boron occurs naturally in sea water at an average concentration of 5 mg B/L and fresh water at 1 mg B/L or less. In dilute aqueous solutions the predominant boron species present is undissociated boric acid. To convert boron oxide into equivalent boron (B) content, multiply by 0.3105.

Phytotoxicity

Boron is an essential micronutrient for healthy growth of plants; however, it can be harmful to boron sensitive plants in higher quantities. Care should be taken to minimize the amount of borate product released to the environment.

Algal toxicity

Green algae, Pseudokirchneriella subcapitata

72-hr EC50 – biomass = 40 mg B/L or 129 mg boron oxide/L [10]

Invertebrate toxicity

Daphnia, Daphnids, *Daphnia magna*

48-hr LC_{50} = 133 mg B/L or 428 mg boron oxide/L [11]

Fish toxicity

Fish, Fathered minnow, *Pimephales promelas*

96-hr LC_{50} = 79.7 mg B/L or 256 mg boron oxide/L [12]

12.2. Persistence and degradability

Boron is naturally occurring and ubiquitous in the environment. Boron oxide decomposes in the environment to natural borate.

12.3. Bioaccumulative potential

Not bioaccumulative.

12.4. Mobility in soil

The product is soluble in water and is leachable through normal soil.

12.5. Results of PBT and vPvB assessment

N.A.

12.6. Other adverse effects

No data available.

SECTION 13. Disposal considerations

13.1. Waste treatment methods

Small quantities of Boroksit can usually be disposed of at landfill sites. No special disposal treatment is required, but local authorities should be consulted about any specific local requirements. Tonnage quantities of product are not recommended to be sent to landfills. Such product should, if possible, be used for an appropriate application.

SECTION 14. Transport information

Boron oxide has no UN Number, and is not regulated under international rail, road, water or air transport regulations.

14.1. UN number **:** N.A.

14.2. UN proper shipping name **:** N.A.

14.3. Transport hazard class(es) **:** N.A.

14.4. Packing group **:** N.A.

14.5. Environmental hazards **:** N.A.

14.6. Special precautions for user **:** N.A.

14.7. Transport in bulk according to Annex II of MARPOL73/78 and the IBC Code **:** N.A.

SECTION 15. Regulatory information

15.1. Safety, health and environmental regulations/legislations specific for the substance

It should be noted that borates are safe under conditions of normal handling and use, besides, they are essential nutrients to plants, and research shows that they play a beneficial role in human health. CLP classification has been solely based on animal tests where animals were exposed to high doses of boric acid over long periods of time. These doses were many times higher than humans are exposed to under conditions of normal handling and use. Consequently, a precautionary decision was taken by the European Commission. Although we will comply with the body of legislation triggered by that decision, we are in process of all possible legal actions.

Clean Air Act (Montreal Protocol)

Boron oxide was not manufactured with and does not contain any Class I or Class II ozone depleting substances.

Chemical inventory listing

- U.S. EPA TSCA : 1303-86-2
- Canadian DSL : 1303-86-2
- EINECS : 215-125-8
- South Korea KECI : KE09919
- South Korea PECs : 231
- Japan ENCS : MITI 1-71, MITI 9-2403
- China IESCS : 1303-86-2
- New Zealand NZIoC : 1303-86-2
- Philippines PICCS : 1303-86-2
- Australia AICS : 1303-86-2

Ensure all national/local regulations are observed.

15.2. Chemical safety assessment

Chemical Safety Assessment of boron oxide (diboron trioxide) has been carried out under REACH Regulation of the EU.

EU Reach Regulation

Diboron trioxide is listed in the Candidate List of Substances of Very High Concern "SVHC" for eventual inclusion in Annex XIV to REACH Regulation 1907/2006 ("Authorisation List"). (15.06.2012-ED/87/2012).

Diboron trioxide is also listed in the Annex XVII of REACH Regulation 1907/2006 (EU No.109/2012) and its use in consumer products above specific concentration limits is restricted. Note that this restriction is only specific to consumer products and do not cover its industrial and/or professional applications. Diboron trioxide can be used in consumer products below specific concentration limits (which is C ≥3.1% for boron oxide).

SECTION 16. Other information

16.1. Mainly changes made to the previous version of this Safety Data Sheet (SDS)

This SDS complies with ISO 11014; the requirements of REACH Title IV and was updated to be in compliance with Annex II of REACH duly amended by **Commission Regulation (EU) No 2015/830 of 28 May 2015**.

Revision No	Revision date	Revision content
07	February 2016	• This SDS was updated in accordance with the ECHA Guidance on the Compilation of Safety data Sheets, Ver. 3.1 dated November 2015.
08	January 2018	• This SDS was updated in line with "Standardization and Simplification of Bag Printings"

16.2. List of abbreviation and acronyms used in this SDS

1st ATP : 1st Adaptation to Technical and scientific Progress
ACGIH : American Conference of Governmental Industrial Hygienists
AICS : Australian Inventory of Chemical Substances
Cal OSHA : The State of California Division of Occupational Safety and Health (DOSH)
Canadian DSL : Canadian Domestic Substances List
CAS N° : Chemical Abstracts Service number
CLP : Classification Labelling Packaging Regulation: Regulation (EC) N°1272/2008
CSR : Chemical Safety Report
DNEL : Derived No effect Level
EC N° : EINECS Number: European Inventory of Existing Commercial Substances
EC_{50} : Half maximal effective concentration
ENCS : Japan Inventory of Existing and New Chemical Substances
Eti Maden : Eti Maden İşletmeleri Genel Müdürlüğü
IECSC : Inventory of Existing Chemical Substances Produced or Imported in China
Index N° : Atomic number of the element most characteristic of the properties of the substance
Japanese MITI : Japanese Ministry of International Trade and Industry
KECI : South Korea Existing Chemicals List
LC_{50} : Lethal Concentration, 50%
LD_{50} : Median Lethal Dose
N.A. : Not Applicable
NZIoC : New Zealand Inventory of Chemicals
OSHA : Occupational Safety & Health Administration
PBT : Persistent, Bioaccumulative and Toxic substance
PECs : South Korea Priority Existing Chemicals

PEL : Permissible Exposure Limits
PICCS : Philippines Inventory of Chemicals and Chemical Substances
PNEC : Predicted No Effect Concentration
REACH : Registration, Evaluation, Authorisation and Restrictions of Chemicals Regulation (EC) N°1907/2006
Repr. Cat. 1B : Substance presumed human reproductive toxicant
SDS : Safety Data Sheet
TLV : Threshold Limit Value
U.S. EPA TSCA : United States Environmental Protection Agency Toxic Substances Control Act
UN : United Nations
vPvB : Very Persistent and Very Bioaccumulative

16.3. List of relevant hazard statements and precautionary statements used in this SDS

According to CLP Regulation
Hazard Statement
H360FD: May damage fertility or the unborn child
Precautionary Statements
Prevention **P201:** Obtain special instructions before use. **P202:** Do not handle until all safety precautions have been read and understood. **P280:** Wear protective gloves/protective clothing/eye protection/face protection. **Response** **P308+P313:** If exposed or concerned: get medical advice/attention. **Storage** **P405:** Store locked up. **Disposal:** **P501:** Dispose of contents/container to in accordance with local regulations.

16.4. Key literature references and sources for data

[1] Fail, P.A., George, J.D., Seely, J.C., Grizzle, T.B., & Heindel, J.J. (1991). Reproductive toxicity of boric acid in Swiss (CD-1) mice: Assessment using the continuous breeding protocol. Fundamental and Applied Toxicology, 17(2), 225-239.

[2] Scialli, A.R., Bonde, J.P., Brüske-Hohlfeld, I., Culver, D.B., Li, Y., & Sullivan, F.M. (2010). An overview of male reproductive studies of boron with an emphasis on studies of highly exposed Chinese workers. Reproductive Toxicology, 29(1), 10-24.

[3] Duydu, Y., Başaran, A., & Bolt, H. (2012). Exposure assessment of boron in Bandırma boric acid production plant. Journal of Trace Elements in Medicine and Biology, 26(2-3), 161-164.

[4] Başaran, N., Duydu, Y., & Bolt, H., (2012). Reproductive toxicity in boron exposed workers in Bandırma, Turkey. Journal of Trace Elements in Medicine and Biology, 26(2-3), 165-167.

[5] Litovitz, T.L., Norman, S.A., & Veltri, J.C. (1986). Annual Report of the American Association of Poison Control Centers National Data Collection System. The American Journal of Emergency Medicine, 4(5), 427-458.

[6] Denton, S.M. (1996). Acute oral toxicity study in the rat: anhydrous boric acid. Final report. Testing laboratory: Corning Hazleton (Europe) Otley Road, Harrogate, North Yorkshire, UK. Report no.: 1341/7-1032. Owner Company: Borax Europe Ltd. Report date: 1996-03-06.

[7] Weiner, A.S., Conine, D.L., & Doyle, R.L. (1982). Acute Dermal Toxicity Screen in Rabbits; Primary Skin Irritation Study in Rabbits of Boric Acid. Testing laboratory: Hill Top Research, Inc. Report no.: 82-0280-21. Owner Company: US Borax Chemical Corporation. Report date: 1982-03-15.

[8] Heindel, J.J., Price, C.J., Field, E.A., Marr, M.C., Myers, C.B., Morrissey, R.E. & Schwetz, B.A. (1992). Developmental toxicity of boric acid in mice and rats. Fundamental and Applied Toxicology, 18(2), 266-277.

[9] Price, C.J., Marr, M.C., Myers, C.B., Heindel, J.J., & Schwetz, B.A. (1991). Final Report on the Developmental Toxicity of Boric Acid (CAS No 10043-35-3) in New Zealand White Rabbits. National Toxicology Program, National Institute of Environmental Health Sciences. Testing laboratory: National Toxicology Program, National Institute of Environmental Health Sciences (TER 90-003; NTIS Accession No PB92-129550). Report no.: TER 90-003; NTIS Accession No PB92-129550.

[10] Hanstveit, A.O. & Oldersma, H. (2000). Determination of the effect of Boric acid, Manufacturing grade on the growth of the fresh water green alga Selenastrum capricornutum. Testing laboratory: TNO Nutrition and Food Research Institute. Report no.: V99.157. Owner Company: Borax Europe Limited. Study number: IMW-99-9047-05. Report date: 2000-03-06.

[11] Gersich, F.M. (1984a). Evaluation of a Static Renewal Chronic Toxicity Test Method for Daphnia magna straus using Boric Acid. Environmental Toxicology and Chemistry, 3(1), 89-94.

[12] Soucek, D., Dickinson, A., & Major, K. (2010). Acute and chronic toxicity of boron to freshwater organisms. Testing laboratory: Illinois Natural History Survey, University of Illinois, Champaign, Illinois. Owner Company: Illinois Natural History Survey, University of Illinois.

For general information on the toxicology of borates see ECETOC Technical Report No. 63 (1995); Patty's Industrial Hygiene and Toxicology, 4th Edition Vol. II, (1994) Chap. 42, 'Boron'.

16.5. Disclaimer of Liability

The information in this SDS was obtained from sources which we believe are reliable. However, the information is provided without any warranty, express or implied, regarding its accuracy, reliability or completeness. The conditions or methods of handling, storage use or disposal of the product are beyond our control and may be beyond our knowledge. For this and other reasons, we do not assume responsibility and expressly disclaim liability for loss, damage or expense arising out of or in any way connected with the handling, storage, use or disposal of the product. It is the user's responsibility to satisfy himself as to the suitableness and completeness of such information for his own particular use.

This SDS was prepared and is to be used only for this product. If the product is used as a component in another product, this SDS information may not be applicable.

Safety Data Sheet Prepared by Arzu DEMİŞ

Certificate Date: 30.09.2015

Certificate Number: 01.58.04

Safety Data Sheet Prepared by Zeynep GÜRTÜRK

Certificate Date: 30.09.2015

Certificate Number: 01.58.07